

Influence of emulsifiers on double emulsion stability

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Agenda

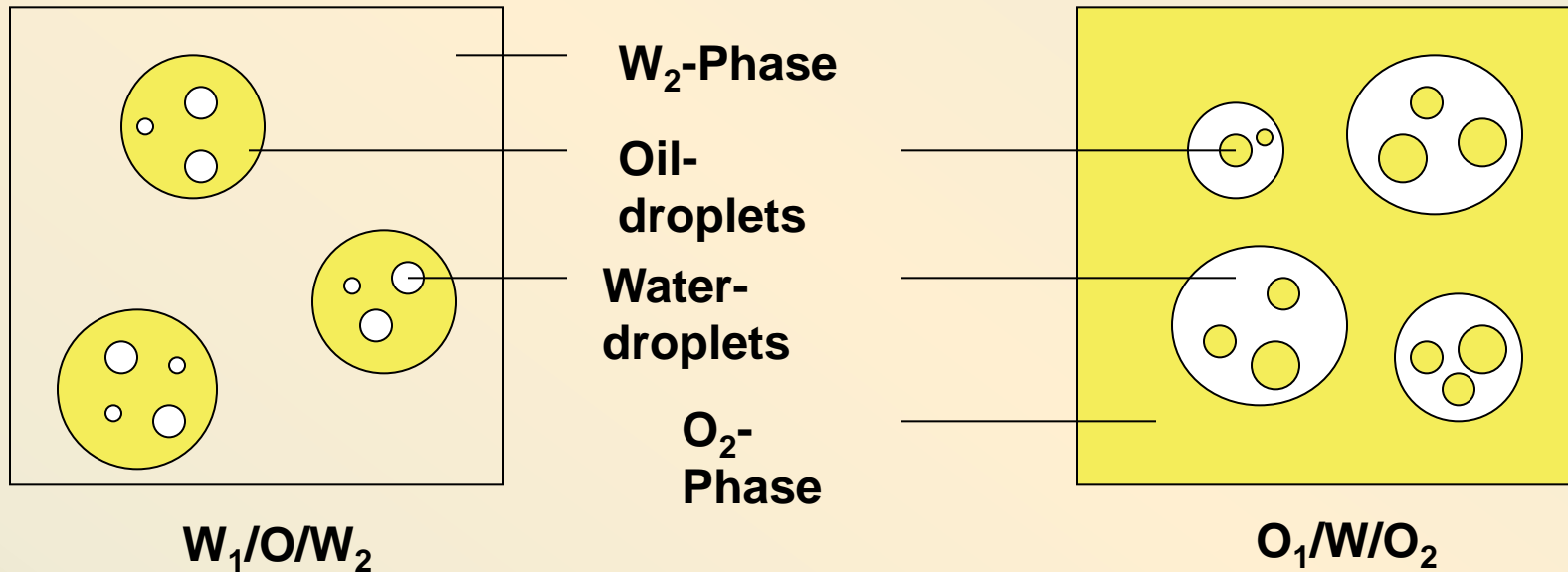
- **What are double (multiple) emulsions?**
- **Emulsion preparation, influence of homogenization method**
- **Influences of emulsifiers**
 - **W/O emulsifier**
 - **O/W emulsifier**
- **Summary**

What are double (multiple) emulsions?

Double emulsions

Definition:

Droplets of a dispersed phase (Oil or Water) contain small droplets of another phase



Special multiple types: $O/W/O/W/O$ and $W/O/W/O/W$

Advantages of double emulsions

- Controlled aroma release
- Encapsulation of bioactive components
- Reduction of fat content without changing the mouthfeeling

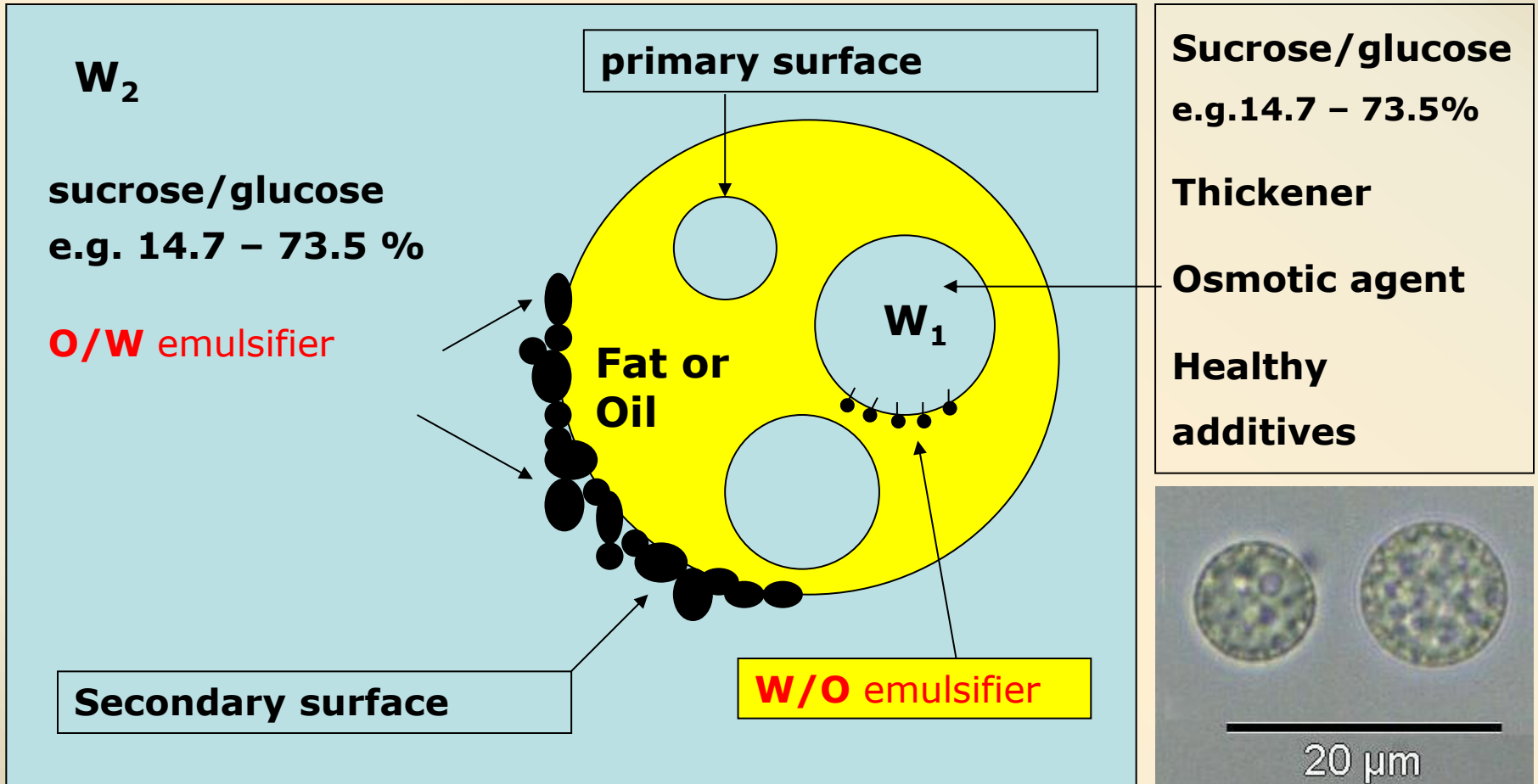
Problem:

- Finding the optimal ingredient composition to
- realize a long time stability

Important:

Selection of emulsifier type and emulsifier combination

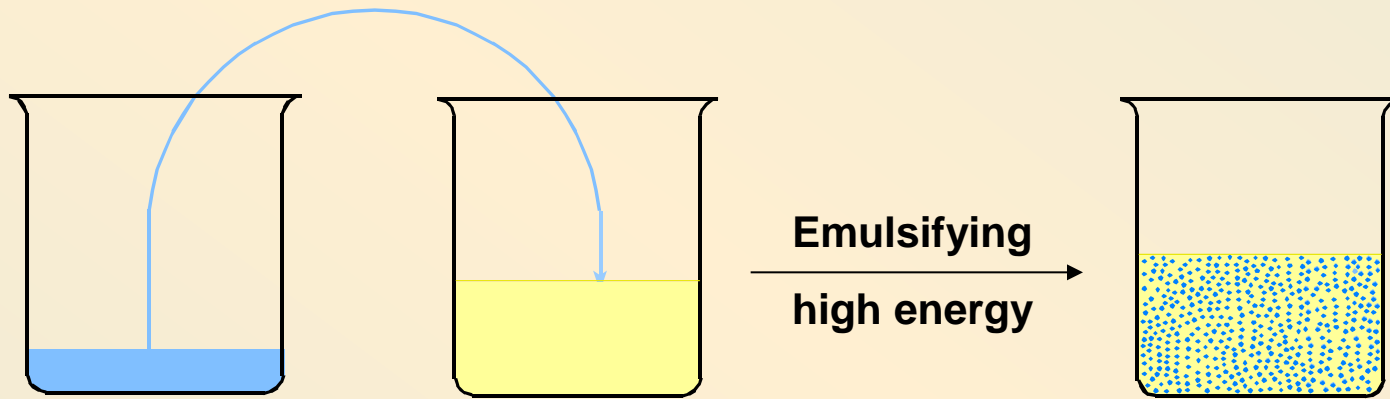
Structure of confectionery W/O/W



Emulsion preparation, influence of homogenization method

Principle of formation multiple emulsions

1. W_1/O - Emulsion



W_1 -phase
water + sugar
and
hydrocolloid,

osmotic
pressure
regulated

Oil-phase +
W/O
emulsifier

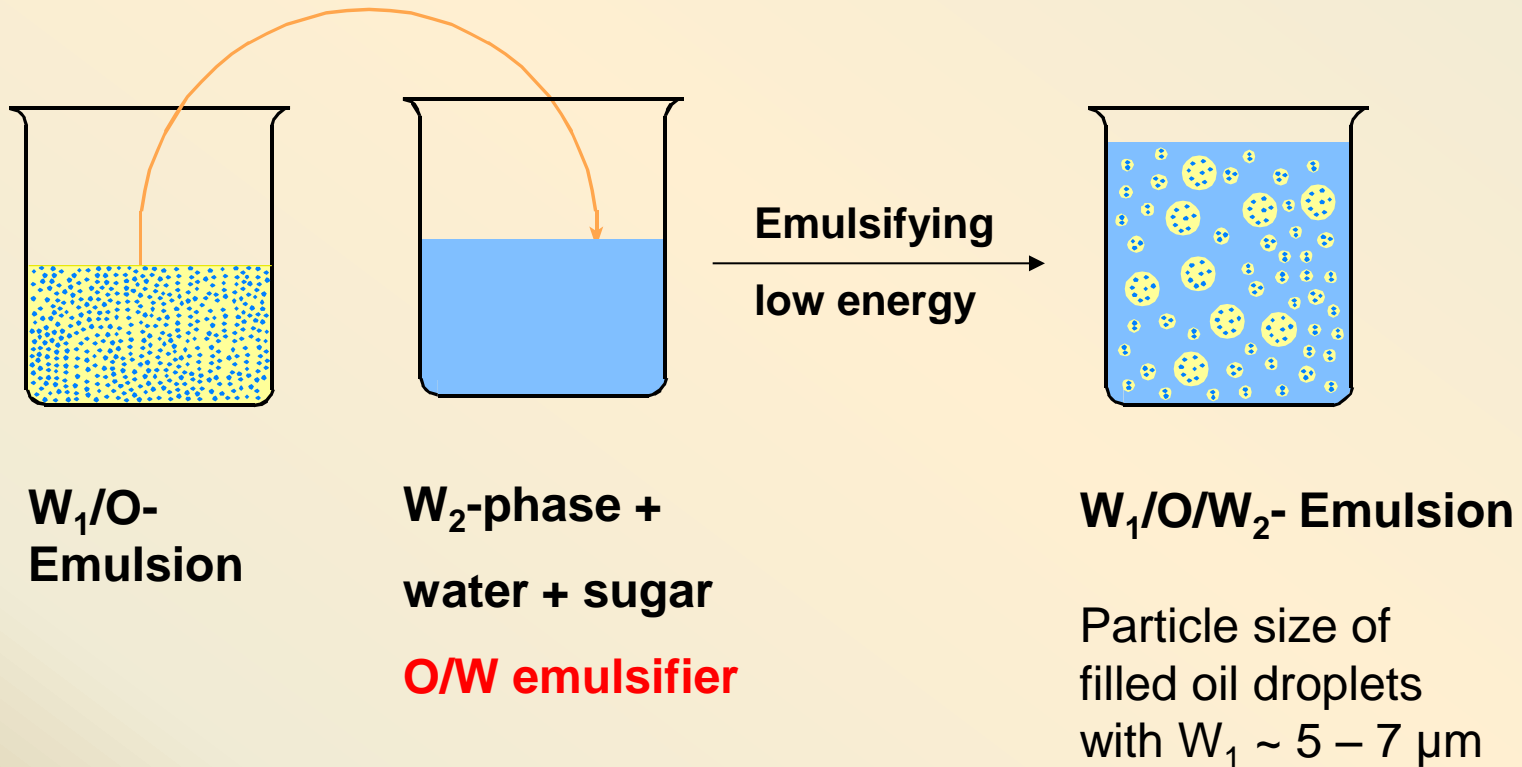
fat
composition
tailored

W_1/O -
emulsion

particle size
of primary
water droplets
~ 1 μm

Principle of multiple emulsion preparation

2. $W_1/O/W_2$ - Emulsion



Methods for emulsion preparation (emulsification of W_1/O in W_2)

Avoid:

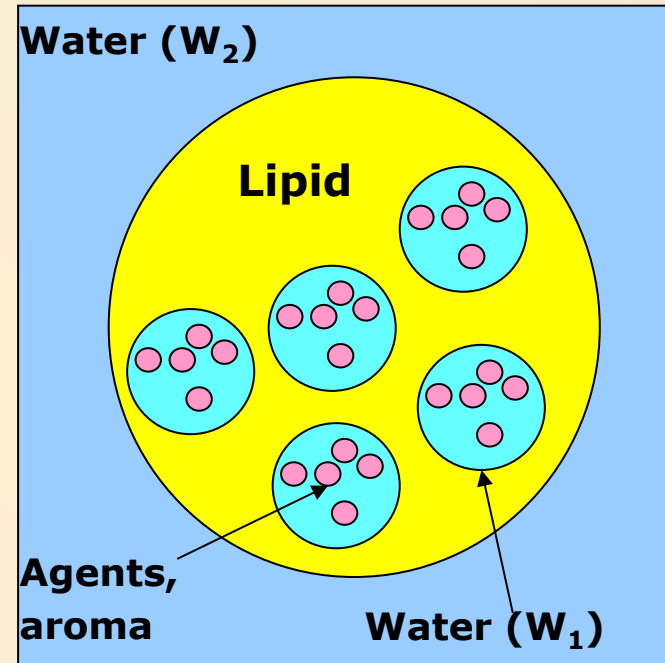
Disruption of internal emulsion droplets (W_1) and fusion with the external phase (W_2)

Emulsification methods for multiple systems are described by:

Muschiolik and Bunjes, 2007 (Behr's Verlag, 2007)

Stability of W/O/W is influenced by:

- Size of inner water droplets (W_1)
- Oil droplet size
- Osmotic gradient between W_1 and W_2
- Laplace curvature pressure
- Water flow between W_1 and W_2 (influenced by osmotic gradient)
- Viscosity of the emulsion phases
- **Interaction between W/O and O/W-emulsifier**



and O/W-emulsifier

Hindrance between different emulsifiers should be prevented!

Influence of emulsifiers

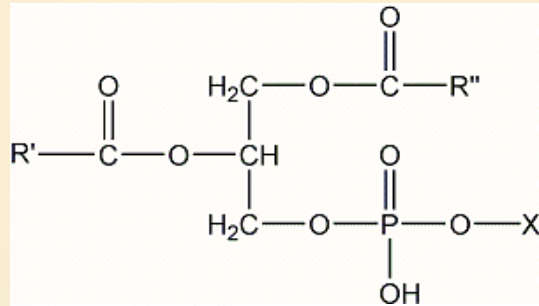
W/O emulsifier lecithin

PC depleted and PE enriched

< 5%

> 11 %

low molecular



Phosphatidyl-

R'-CO; R''-

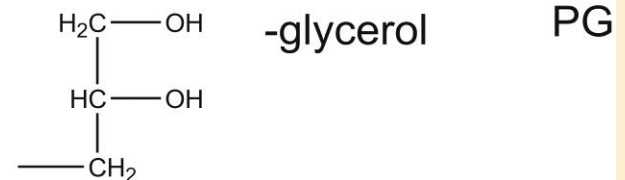
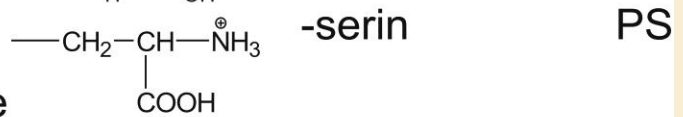
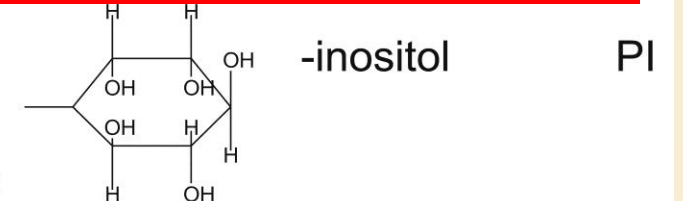
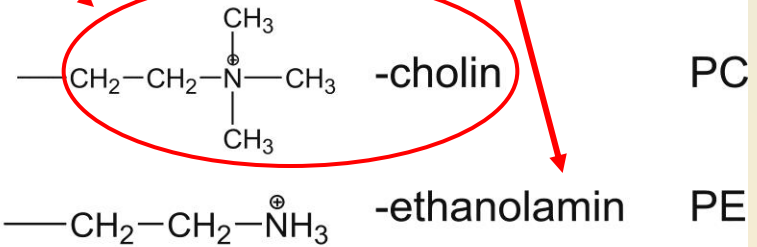
CO = Fettsäurereste

X = H: Phosphatidsäuren

R'-CO; R''- (PA)

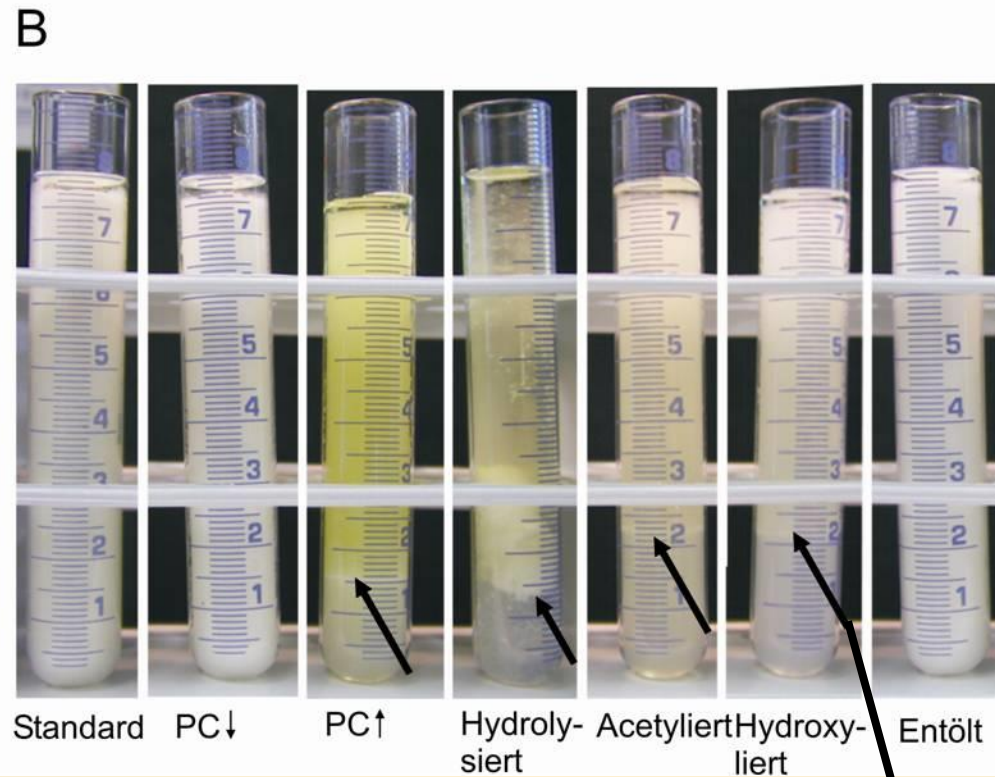
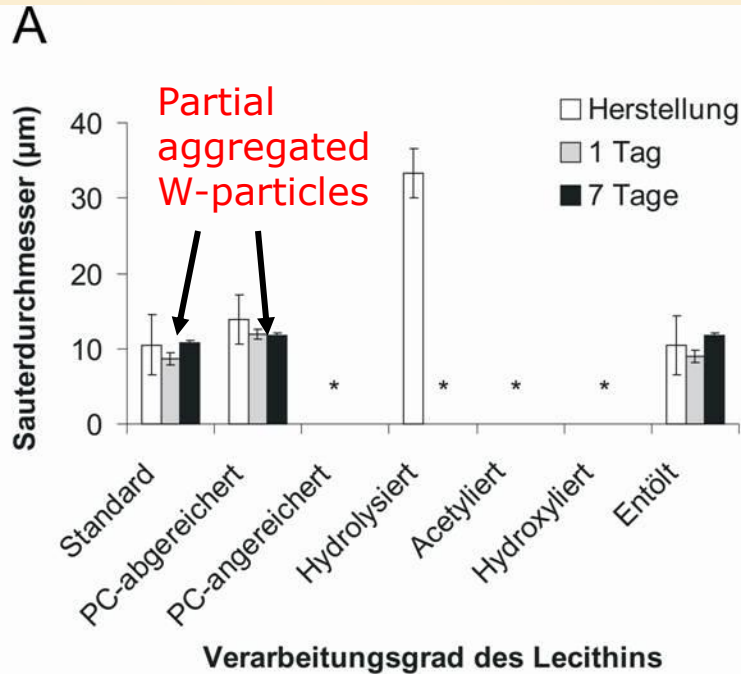
CO = H: Lysophospholipide

X=



Hydrophobic lecithin with higher PE content as W/O emulsifier

W/O-Premix prepared with different lecithins



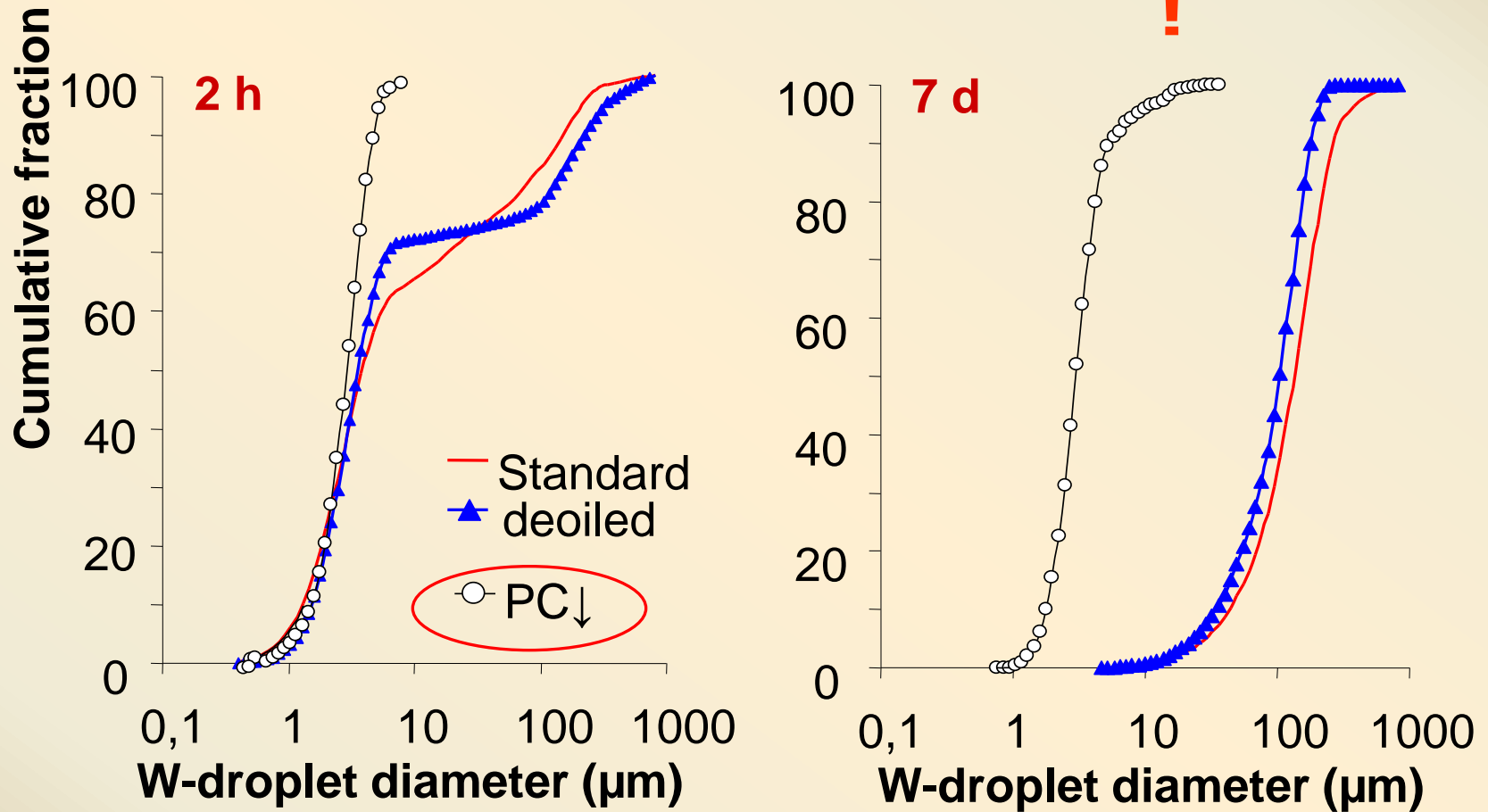
Improved W/O stability with higher PE content

*Phase separation

W/O : 30/70; O-phase: sunflower oil, emulsion preparation at 3000 rpm, 2 min;
 @ 50 °C; Homogenizer MPW-302 (Metronex, Poland)

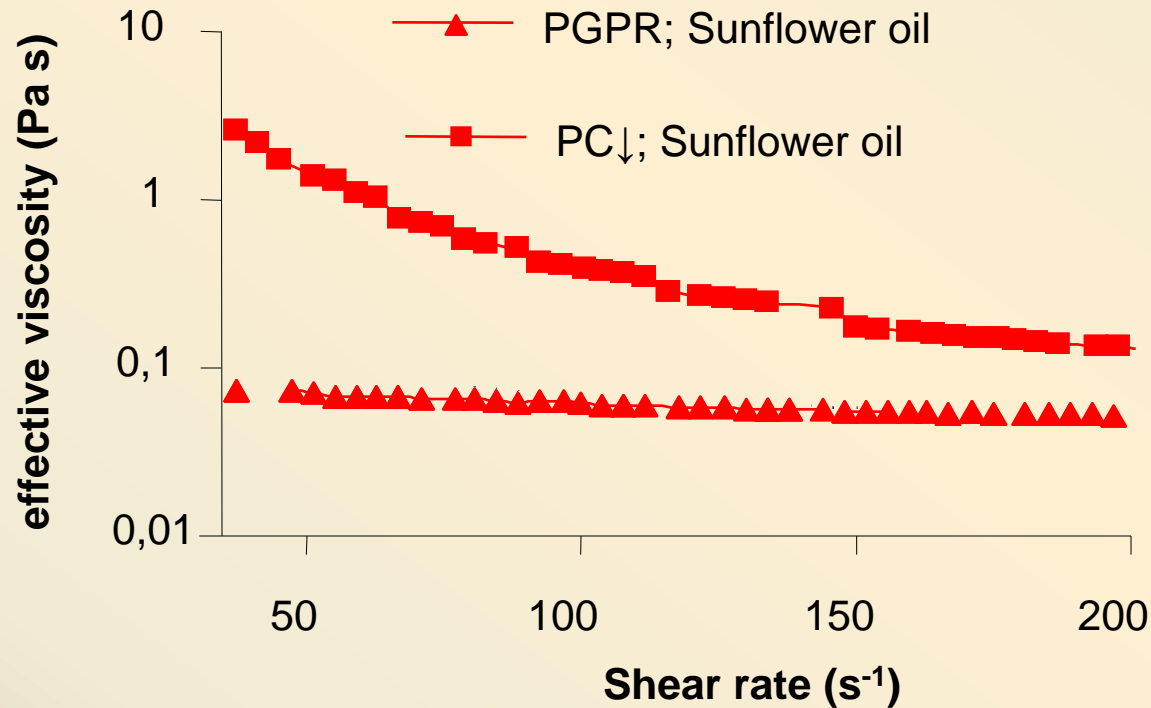
Lecithin-screening

- W/O emulsions with 0.75 % Lecithin (2.5 % in O)



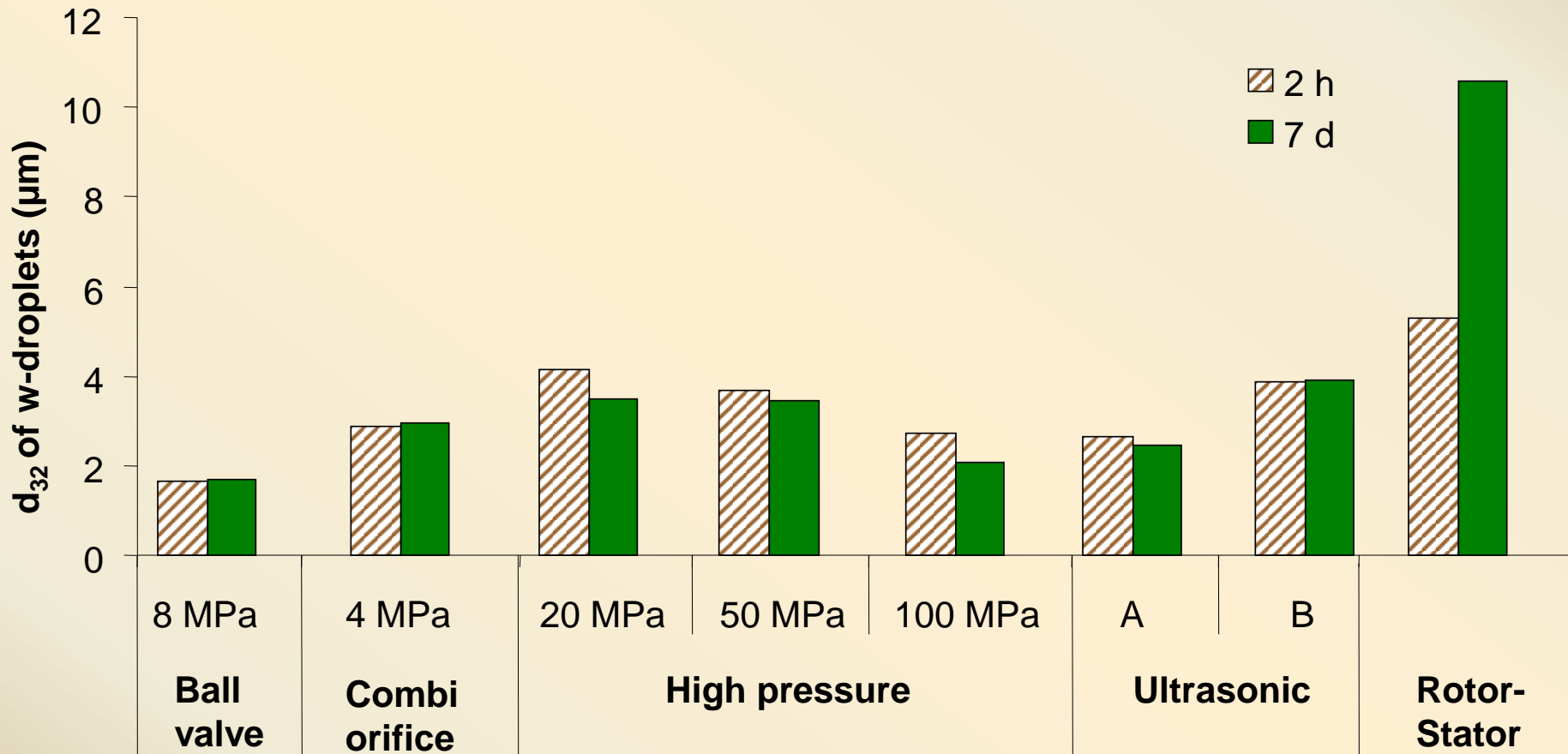
Cumulative particle-size distribution for W-droplets

Flow behaviour of W/O-Emulsionen prepared with lecithin



Due to aggregation of water particles W/O emulsions with lecithin are higher viscous

Influences of emulsification methods - W/O with lecithin -



A: 10 ml, 240 s ultrasonic, 60 % amplitude; B: 30 ml, 240 s ultrasonic, 60 % amplitude

0.2 % XPS and 1.5 % whey protein in W; 2.5 % PC↓ in O; 50 °C

W/O with lecithin

W/O phases prepared with lecithin

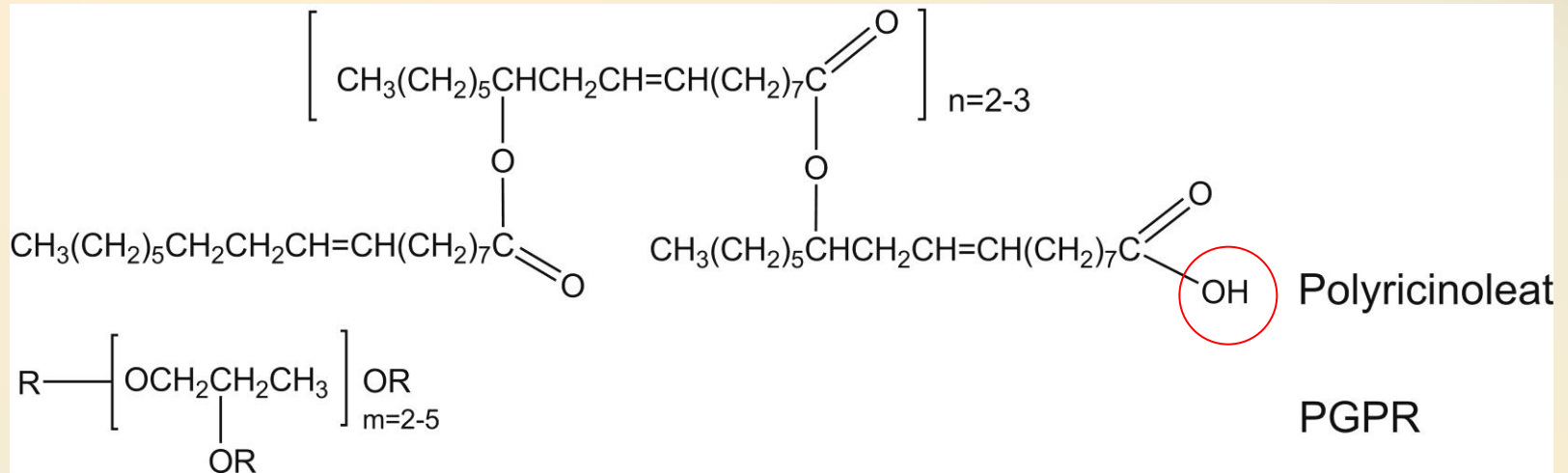
- are more viscous than w/o phases with PGPR

and

- the stability depends strongly on the emulsifying methods

W/O emulsifier PGPR

Polyglycerol-Polyricinoleate E476

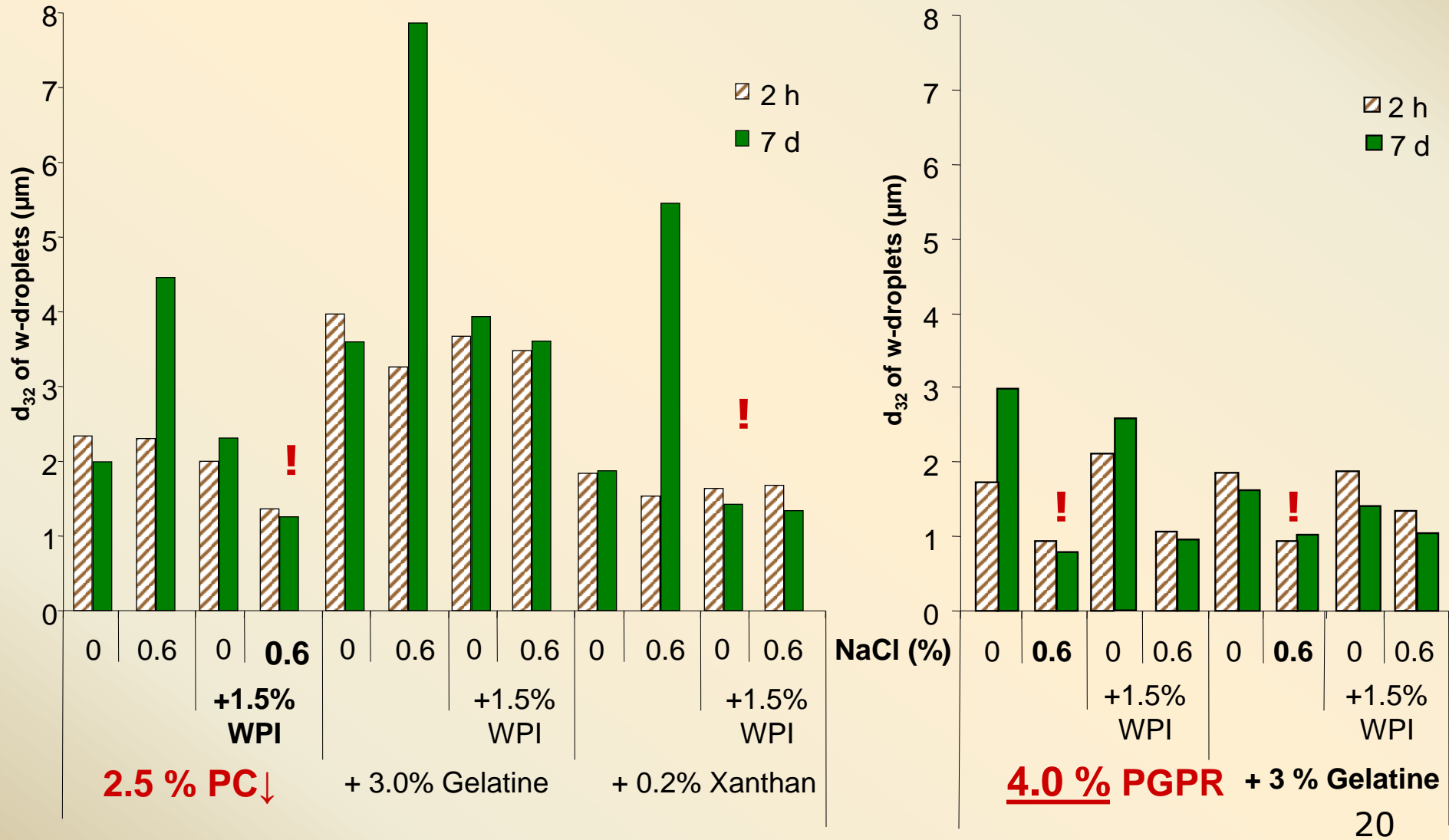


ZZuIV, 29.01.98, appendix 7

- max. 4 g/kg in spreads with less than 41 % fat, spreads with < 10 % fat and salat dressings
- max. 5 g/kg in sweets with cacao

Lecithin or PGPR in O

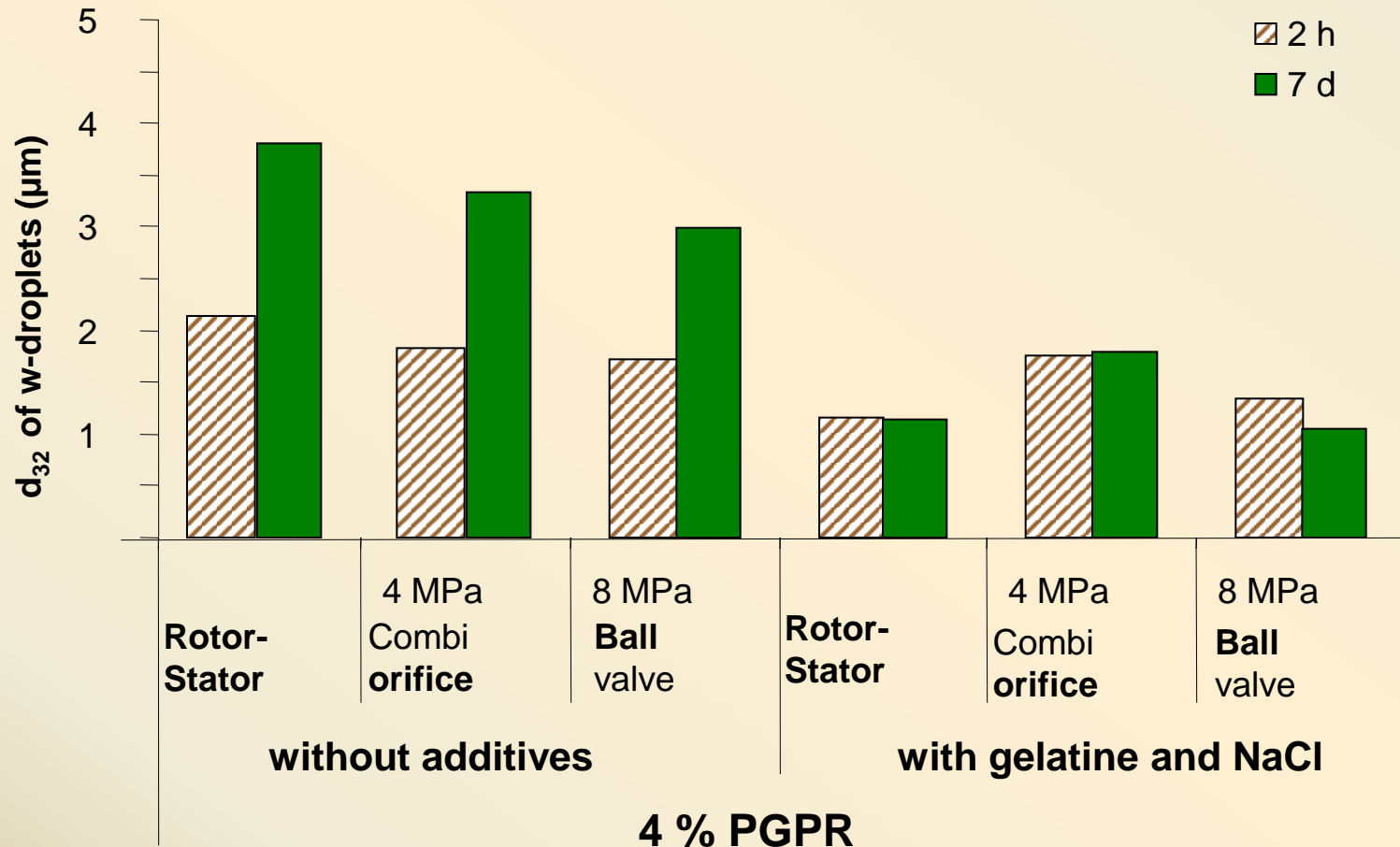
- Influence of NaCl and WPI in W_1 -phase -



Influence of other emulsion components:

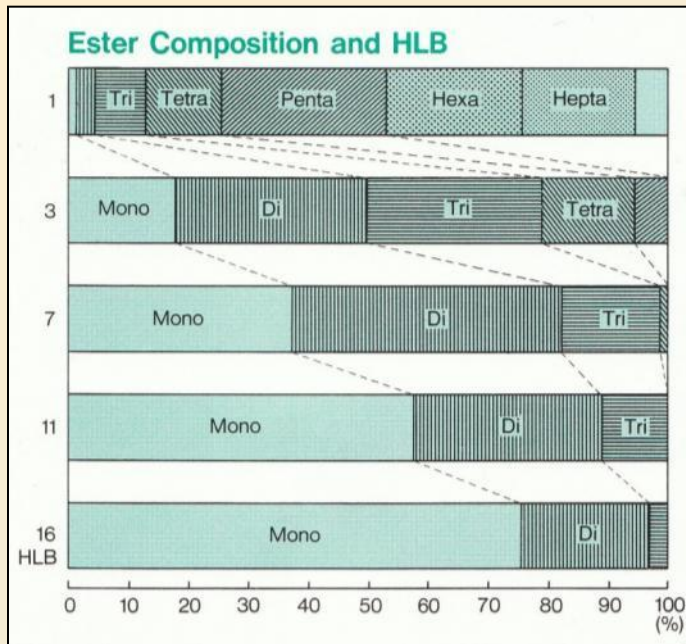
- Electrolytes (NaCl) are essential to achieve coalescence-stable emulsions prepared with PGPR.
- Electrolytes (NaCl) in emulsions containing lecithin contribute more to coalescence of water droplets and phase separation (other components are necessary to regulate the osmotic pressure, e.g. glucose).
- Combination of lower surface active whey protein with xanthan in W-phase (1.5 % protein + 0.2 % XPS in W) reduces the W-droplets additionally when W/O emulsions are prepared with PC depleted lecithin (2.5 % in O).

Influence of emulsification method - W/O with PGPR -

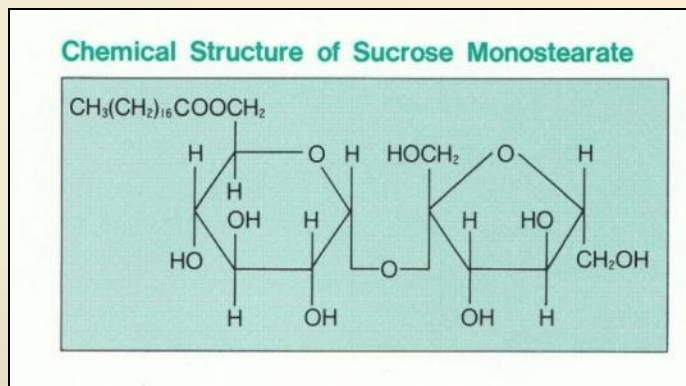


4 % PGPR
Rotor-stator-system is effective!

W/O emulsifier sucrose esters



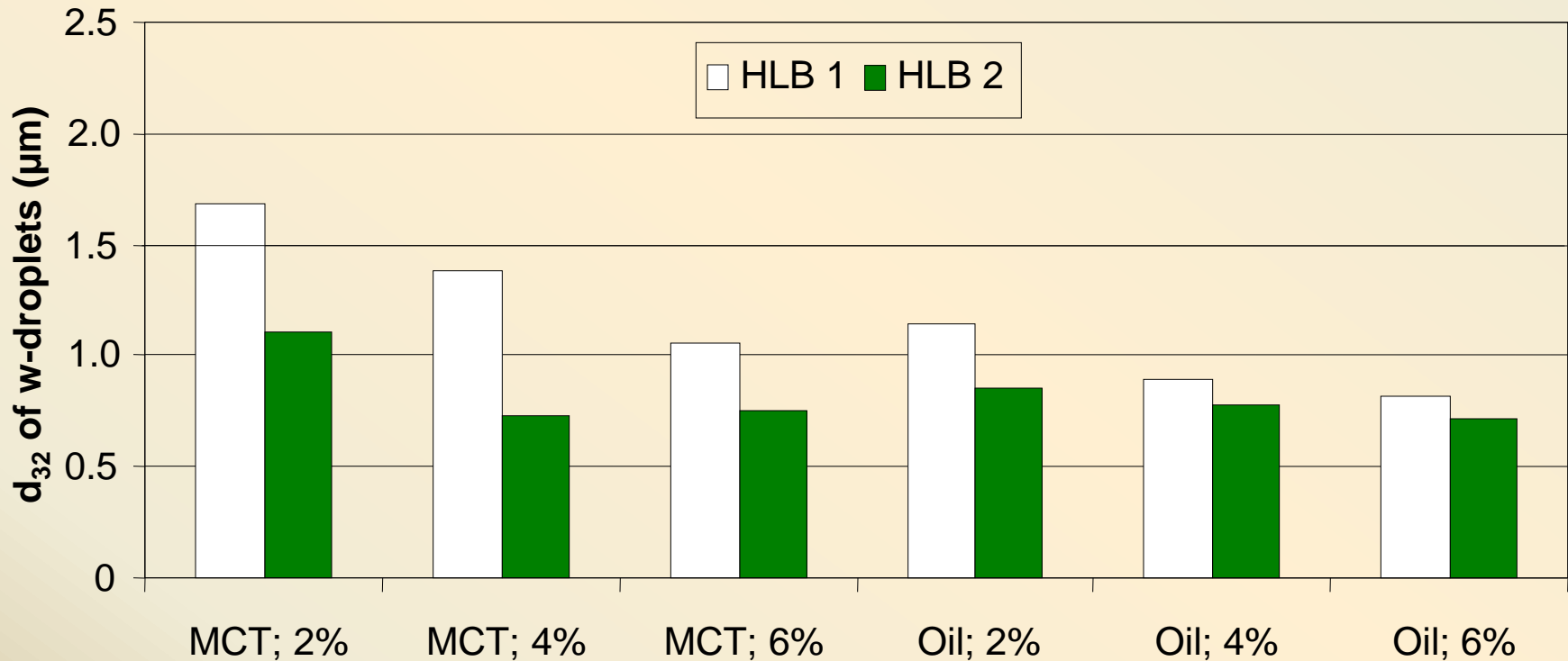
- **Sucrose erucate ER-190, HLB 1**
- **Sucrose erucate ER-290, HLB 2**
- **Sucrose esters, which are allowed in the European Union have to contain at least 80 % mono-, di-, and tri-esters**
- **At the present time the tested sucrose esters with HLB 1 and 2 are not permitted**



Particle size of W/O

Sugar esters with HLB 1 and HLB 2

(2, 4 and 6 % emulsifier in O; 0.6 % **NaCl** in W)

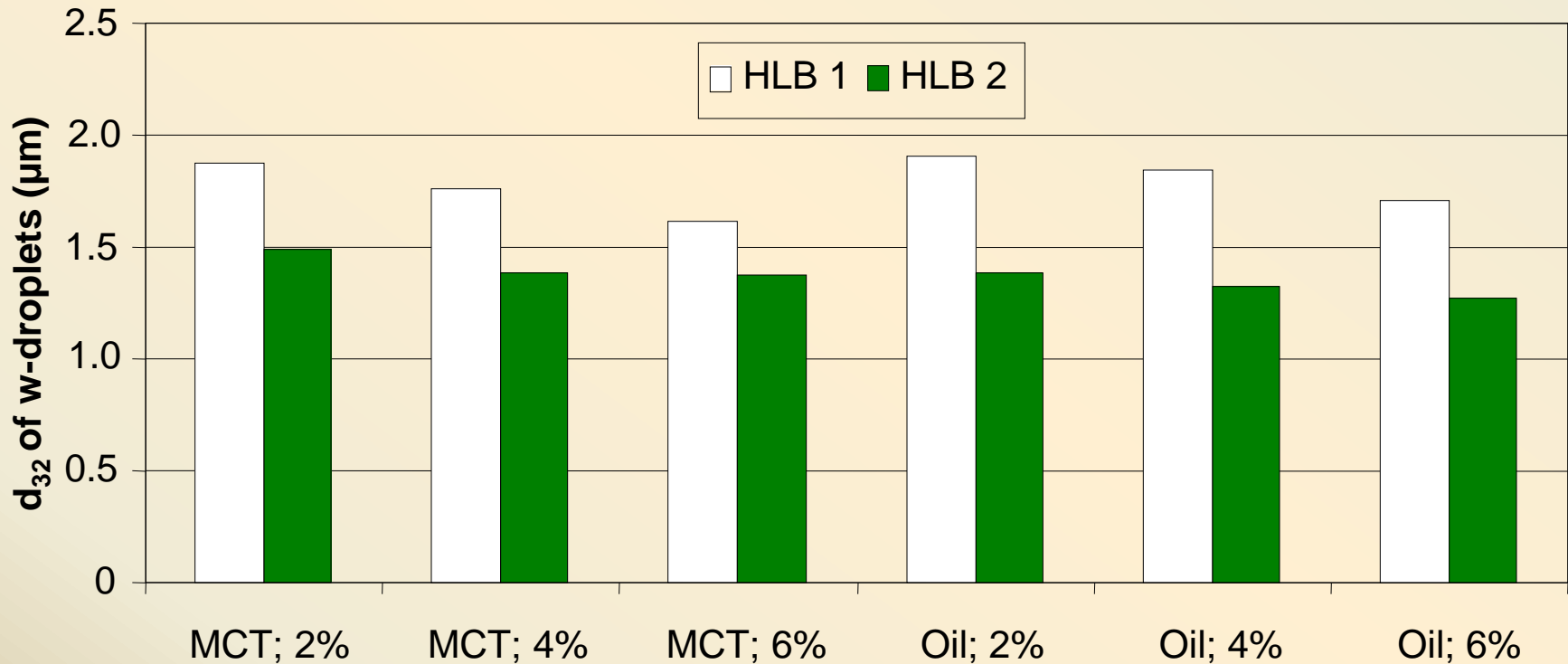


MCT: Medium-chain triglyceride (Miglyol 812), Oil: Vegetable oil (BISKIN)

Particle size of W/O

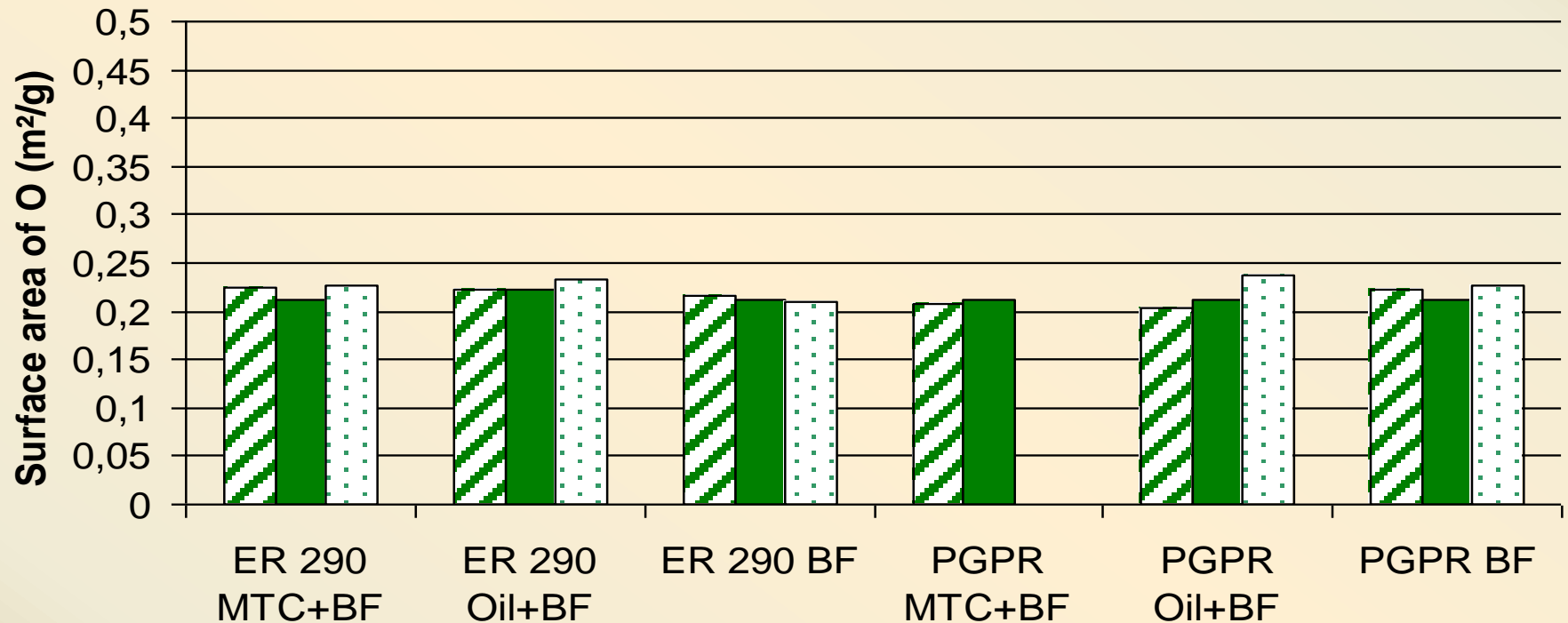
Sugar ester with HLB 1 and HLB 2

(2, 4 and 6 % emulsifier in O)



Surface area of O-droplets (W/O in W) depending on storage (Particle size between 26 – 29 μm)

▨ 1 day ■ 7 days □ 35 days



W₂: 25 % dry matter (sugar, milk protein, starch, maltodextrin)

4 % sugar ester (ER 290, HLB 2); PGPR: 4 %;

BF: butter fat; MCT: Miglyol 812; Oil: vegetable oil

W₁/O = 20:80, (W₁/O) : W₂ = 20:80

Summarised effects of sugar esters

- Esters with HLB 2 are more effective in forming small W_1 particles than esters with HLB 1
- Application of NaCl in W_1 supports the formation of small water droplets
- W/O are stable with 4 % sugar ester in O (HLB 2) and 0.6 % NaCl in W_1
- The size of O -droplets (surface area) in $W/O/W$ is comparable to emulsions with PGPR

Emulsifiers for W_1/O in W_2

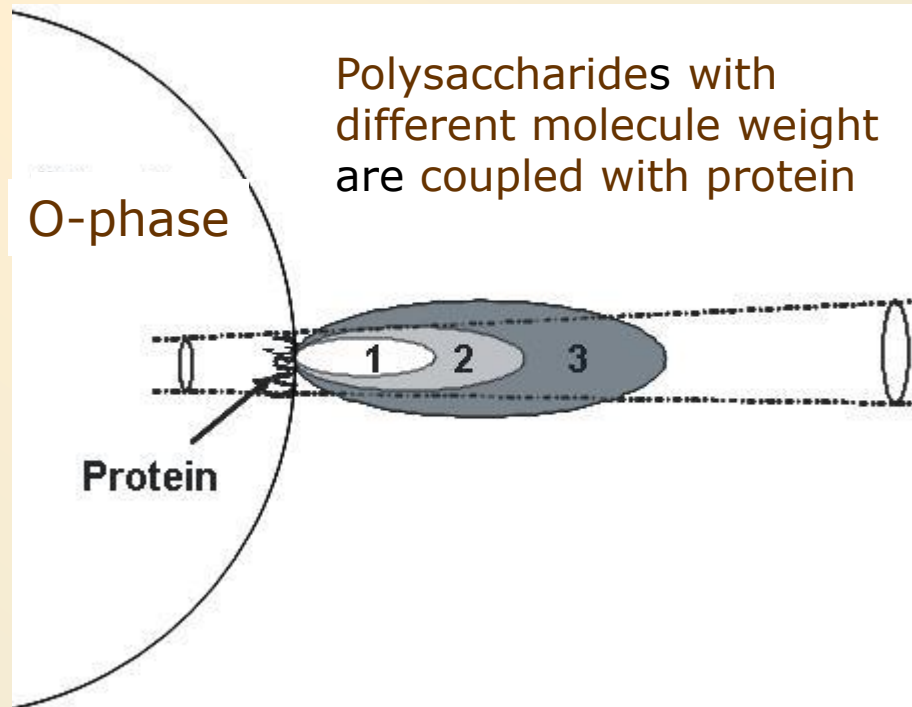
Emphasized O/W emulsifiers for double emulsions

- Native proteins (whey, vegetable)
- Protein-ionic polysaccharide mixtures
(a high zeta-potential is advantageous)
- Protein-polysaccharide conjugates

Combination of proteins with ionic polysaccharides
in W_2 increases the barrier function!

Structural compatibility between
W/O- and O/W-emulsifiers is of importance!

Protein-polysaccharide conjugate at O/W-surfaces

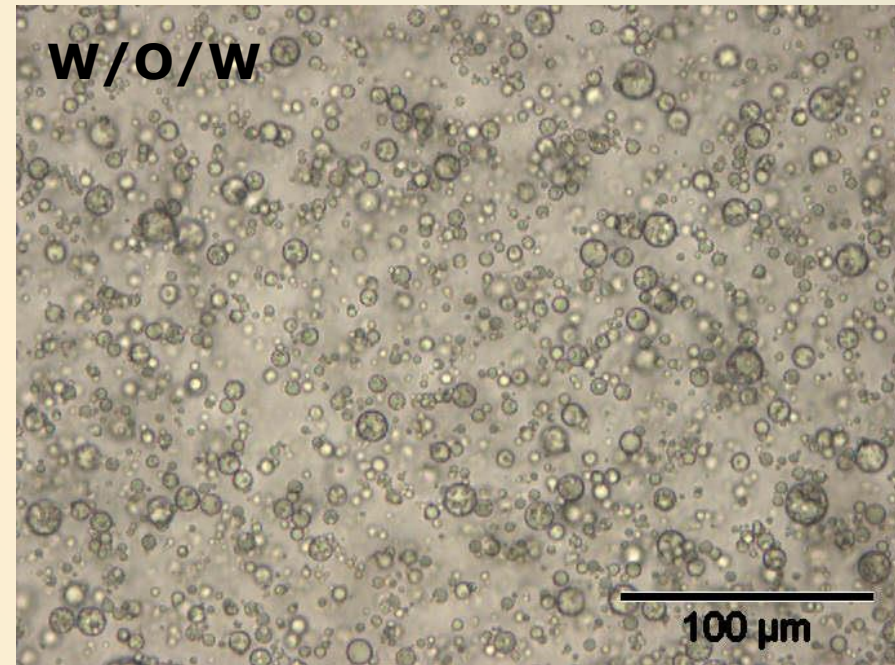
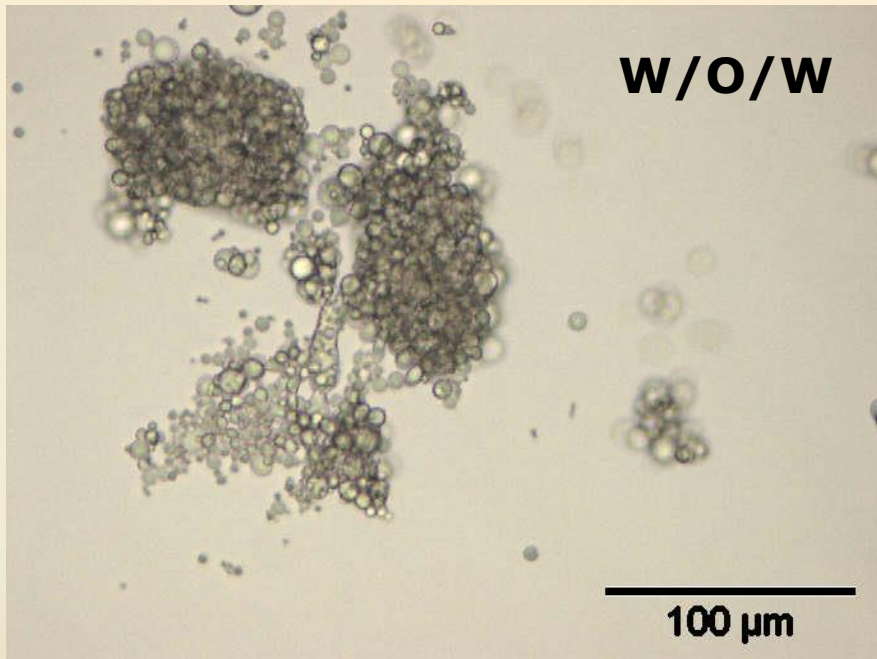


according to Dunlap u. Côté, 2005

Available space for adsorbing at oil surfaces depends on the polysaccharides molecule weight

Milk protein and protein-pectin-conjugate as O/W-emulsifier in W/O/W

W/O/W were heated for 10 min at 90 °C, and stored for 6 weeks



Milk protein concentrate in W₂

Conjugate in W₂

Double emulsions with conjugate in W₂ are heat stable!

No aggregation of O-droplets! 31

Influence of O-phase on W/O/W stability

The **stability** can be improved by using O-phases with a high degree of saturated fatty acids, i.e. by fat phases with a low polarity

(promotion of a close packed condensed interfacial film)

Conclusions

Stable W/O/W can be prepared by using

W/O emulsifiers in O:

- 4 % PGPR (0.6 % NaCl in W_1) or
- 2.5 - 6 % PE enriched lecithin (without NaCl) or
- (4 % sucrose ester; HLB 2; 0.6 % NaCl)

O/W emulsifiers in W_2 :

- Whey or vegetable protein or
- Protein-ionic polysaccharide conjugates

Low molecular weight W/O emulsifier

+ high molecular weight O/W emulsifier

= better compatibility and no negative interactions!

Conclusions

The stability of multiple systems depends on:

- osmotic balance, electrolyte and ionic status and
- fat phase composition (saturation degree, polarity)

The selection of emulsifiers for double emulsions has to consider the electrolyte and ionic status.

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Project AiF 14087 BG

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